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## Preparation of highly functionalized arylmagnesium reagents by the addition of magnesium phenylselenide to arynes

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Abstract—The described polyfunctional arylmagnesium reagents, resulting from the highly regioselective addition of magnesium phenylselenide to functionalized arynes, can be trapped by a range of electrophiles, yielding polyfunctional selenoethers in 45–85% yields. Furthermore, these Grignard reagents can be used in Negishi cross-coupling reactions with iodoarenes after transmetalation to the corresponding arylzinc compounds, furnishing functionalized biaryl products in 55–73% yields. 2006 Elsevier Ltd. All rights reserved.

Arynes are highly reactive intermediates, which have found numerous applications in organic synthesis.<sup>[1](#page-2-0)</sup> Of special interest is the addition of nucleophiles to arynes.[1–3](#page-2-0) Recently, we have described a new preparation of polyfunctional arynes by the elimination of 2-magnesiated diaryl sulfonates prepared from the corresponding iodides of type  $1^{\tilde{A}}$  Application of this method enabled the preparation of functionalized arylmagnesium reagents 2 and 3 by the addition of magne-sium arylthiolates 4 and amides [5](#page-3-0) to aryne.<sup>5</sup> In contrast to previous methods, these aryl magnesium reagents can be trapped by electrophiles giving rise to aryl thioethers of type 6 and aryl amines of type 7 (Scheme  $1$ ).<sup>[5](#page-3-0)</sup>

Arylselenoethers are very useful compounds, which can be converted to various compounds.<sup>[6](#page-3-0)</sup> Two examples have been reported for the addition of magnesium phenylselenide (8) to benzyne, followed by trapping with electrophiles.<sup>[5](#page-3-0)</sup> Herein, we report the regioselective addition of magnesium phenylselenide (8) to arynes generated by our previously reported procedure, providing 2-seleno-substituted aryl magnesium species of type 9. The resulting arylmagnesium reagents 9 can be trapped by a range of electrophiles, yielding aryl selenoethers of type 10 ([Scheme 2](#page-1-0) and [Table 1\)](#page-1-0).

Thus, the addition of  $i$ -PrMgCl (2.0 equiv) to phenylselenol (1.0 equiv) in THF  $(-78 \degree C, 0.5 \text{ h})$  followed by



Scheme 1. Preparation of aryl thioethers and aryl amines by addition reactions to benzyne.

Keywords: Aryne; Organomagnesium reagent; Selenoether.

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Scheme 2. Preparation of aryl selenoethers by addition reactions to arynes.

Table 1. Synthesis of selenoethers of type 10 by the addition of magnesium phenylselenide 8 to arynes followed by the trapping of the intermediate Grignard reagents 9 with an electrophile (see Scheme 2)

Entry	$1 \quad$	Electrophile	Product of type $10$ Yield $(\%)^a$	
			R PhSe OCH <sub>3</sub>	
$\mathbf{1}$	1 <sub>b</sub>	$H_2O$	10a: $R = H$	85
		DMF	10 $b: R = CHO$	70
$\begin{array}{c} 2 \\ 3 \\ 4 \end{array}$		Allyl bromide <sup>b</sup>	10c: $R =$ allyl	82
		PhCOCl <sup>e</sup>	10d: $R = COPh$	77
			R PhSe OBn	
5	1c	H <sub>2</sub> O	10e: $R = H$	84
6		DMF	10f: $R = CHO$	72
$\sqrt{ }$		Allyl bromide <sup>b</sup>	10g: $R =$ allyl	74
8		PhCOCl <sup>e</sup>	10 $h: R = COPh$	76
			R PhSe <b>OTES</b>	
9	1d	I <sub>2</sub>	10i: $R = I$	51
10		Allyl bromide <sup>b</sup>	10 <i>j</i> : $R =$ allyl	45

<sup>a</sup> Yield of analytically pure isolated product.

 $\rm^b$  The reaction was performed with 0.5 equiv CuCN 2LiCl.  $\rm^c$  The reaction was performed with 1.0 equiv CuCN 2LiCl.

the addition of 2-iodo-3-methoxyphenyl 4-chlorobenzenesulfonate (1b)  $(1.0 \text{ equiv}; -78 \text{ °C}, 0.5 \text{ h})$  and subsequent stirring at ambient temperature for 1 h led to the aryne-addition product 9b, which upon quenching with water provided 1-methoxy-3-(phenylseleno)-benzene (10a) in 85% isolated yield (Table 1, entry 1). Similarly, intermediate Grignard reagent 9b was formylated with DMF (1.5 equiv,  $-40$  °C to rt, 1 h), leading to 2-methoxy-6-(phenylseleno)-benzaldehyde (10b) in 70% yield (entry 2).[7,8](#page-3-0) Allyl bromide and benzoyl chloride  $(in the presence of CuCN·2LiCl)$  also served as excellent electrophiles, yielding 10c and 10d in 82% and 77%, respectively (entries 3 and 4).

The intermediate Grignard reagent 9c, which was formed from 2-iodo-3-benzyloxyphenyl 4-chlorobenzenesulfonate (1c) under the same reaction conditions as 9b, was successfully quenched with a range of electrophiles to give rise to the corresponding products 10e (84% yield), 10f (72% yield), 10g (74% yield), and 10h (76% yield), respectively (entries 5–8).

Even the more sterically hindered Grignard reagent 9d generated from 2-iodo-3-triethylsilanyloxyphenyl 4 chlorobenzenesulfonate (1d), was successfully trapped with iodine (1.5 equiv,  $-78$  °C to rt, 1 h) to give rise to 10i in 51% yield (entry 9). Furthermore, the reaction of 9d with allyl bromide gave rise to the allylated selenoether 10j in 45% yield (entry 10).

Interestingly, functionalized arynes displayed a remarkable regioselectivity in the addition step. $5$  Thus the polyfunctional sulfonate 11 was selectively magnesiated at the ortho-position of the sulfonate group, and its reaction with magnesium phenylselenide (8) provided selec-



Scheme 3. Reagents and conditions: (a) THF, PhSeMgCl  $(2.0 \text{ equiv})$ ,  $i$ -PrMgCl  $(1.0 \text{ equiv})$ ,  $-78 \text{ °C}$ , 0.5 h; rt, 2 h; (b) CuCN<sup>2</sup>LiCl  $(1.0 \text{ equiv})$ ,  $-78$  °C, 10 min; EtCOCl (3.0 equiv),  $-78$  °C to rt, 1 h; (c) CuCN·2LiCl (0.5 equiv),  $-78$  °C, 10 min; allyl bromide (3.0 equiv),  $-78$  °C to rt, 1 h.

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**Scheme 4.** Reagents and conditions: (a) THF, PhSeMgCl (1.0 equiv), *i*-PrMgCl (1.0 equiv),  $-78$  °C, 0.5 h; 0 °C, 10 min; (b) ZnBr<sub>2</sub> (1.0 equiv),  $-78$  °C, 10 min; iodoarenes (1.5 equiv), Pd(dba)<sub>2</sub> (5 mol %), tfp (10 mol %),  $-40$  °C to rt, 5 h.

tively the magnesiated reagent 12, which was stabilized by chelation. Its reaction with various electrophiles, such as an acid chloride, or allyl bromide in the presence of CuCN.2LiCl, furnished the tetrasubstituted selenoethers  $13a (60\% \text{ yield})^8$  $13a (60\% \text{ yield})^8$  and  $13b (64\% \text{ yield})$ , respectively ([Scheme 3\)](#page-1-0).

Furthermore, the arylmagnesium reagent 14, which resulted from the reaction of 8 with 1a under similar reaction condition, can be subjected to Negishi crosscoupling reaction. After transmetalation with  $ZnBr<sub>2</sub>$ , the corresponding zinc reagents can react with various iodoarenes 15a–c under standard reaction conditions, furnishing functionalized biaryls  $16a-c^8$  $16a-c^8$  in 55–73% yields (Scheme 4 and Table 2).

In summary, we have developed a general procedure for the selenomagnesiation of arynes. The resulting functionalized arylmagnesium species can be trapped with numerous electrophiles including iodoarenes, which undergo smooth Negishi cross-coupling reactions, in contrast to most previously reported addition reactions.





<sup>a</sup> Yield of analytically pure isolated product.

Further extensions of this work, utilizing other heteroarynes, are currently underway in our laboratories.

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## References and notes

- 1. For excellent reviews see: (a) Sauer, J.; Huisgen, R. Angew. Chem. 1960, 72, 294; (b) Hoffmann, R. W. Dehydrobenzene and Cycloalkynes; Academic Press: New York, 1967; (c) Castedo, L.; Guitian, E. In Studies in Natural Products Chemistry; Atta-ur Rahman, Ed.; Elsevier: Amsterdam, 1989; Vol. 3, Part B, p 417; (d) Kessar, S. V. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Semmelhack, H. F., Eds.; Pergamon Press, 1991; Vol. 4, p 483; (e) Brown, R. F. C.; Eastwood, F. W. Synlett 1993, 9; (f) Sander, W. Acc. Chem. Res. 1999, 32, 669; (g) Pelissier, H.; Santelli, M. Tetrahedron 2003, 59, 701.
- 2. (a) Hamura, T.; Ibusuki, Y.; Sato, K.; Matsumoto, T.; Osamura, Y.; Suzuki, K. Org. Lett. 2003, 5, 3551; (b) Hosoya, T.; Hamura, T.; Kuriyama, Y.; Miyamaoto, M.; Matsumoto, T.; Suzuki, K. Synlett 2000, 520; (c) Hosoya, T.; Takashiro, E.; Matsumoto, T.; Suzuki, K. J. Am. Chem. Soc. 1994, 116, 1004; (d) Beller, M.; Breindl, C.; Riermeier, T. H.; Tillack, A. J. Org. Chem. 2001, 66, 1403; (e) Maurin, P.; Ibrahim-Ouali, M.; Santelli, M. Tetrahedron Lett. 2001, 42, 8147; (f) Liu, Z.; Larock, R. C. Org. Lett. 2003, 5, 4673; (g) Yoshida, H.; Honda, Y.; Shirakawa, E.; Hiyama, T. Chem. Commun. 2001, 1880; (h) Yoshida, H.; Shirakawa, E.; Honda, Y.; Hiyama, T. Angew. Chem. 2002, 114, 3381; Angew. Chem., Int. Ed. 2002, 41, 3247; (i) Peña, D.; Escudero, S.; Pérez, D.; Guitián, E.; Castedo, L. Angew. Chem. 1998, 110, 2804; Angew. Chem., Int. Ed. 1998, 37, 2659; (j) Becht, J.-M.; Gissot, A.; Wagner, A.; Mioskowski, C. Chem. Eur. J. 2003, 9, 3209; (k) Liu, Z.; Larock, R. C. Org. Lett. 2004, 6, 99; (l) Yoshida, H.; Watanabe, M.; Ohshita, J.; Kunai, A. Chem. Commun. 2005, 3292; (m) Zhao, J.; Larock, R. C. Org. Lett. 2005, 7, 4273; (n) Liu, Z.; Larock, R. C. J. Am. Chem. Soc. 2005, 127, 13112; (o) Tambar, U. K.; Stoltz, B. M. J. Am. Chem. Soc. 2005, 127, 5340; (p) Yoshida, H.; Minabe, T.; Ohshita, J.; Kunai, A. Chem. Commun. 2005, 3454; (q) Yoshida, H.; Minabe, T.; Ohshita, J.; Kunai, A. Tetrahedron Lett. 2005, 46, 6729.
- 3. (a) Jamart-Gregoire, B.; Leger, C.; Caubère, P. Tetrahedron Lett. 1990, 31, 7599; (b) Tripathy, S.; LeBlanc, R.; Durst, T. Org. Lett. 1999, 1, 1973; (c) Paz, M.; Saa, C.; Guitian, E.; Castedo, L.; Saa, J. M. Heterocycles 1993, 36, 1217;

<span id="page-3-0"></span>(d) Fleming, I.; Woolias, M. J. Chem. Soc., Perkin Trans. 1 1979, 827.

- 4. Sapountzis, I.; Lin, W.; Fischer, M.; Knochel, P. Angew. Chem. 2004, 116, 4464; Angew. Chem., Int. Ed. 2004, 43, 4364.
- 5. Lin, W.; Sapountzis, I.; Knochel, P. Angew. Chem. 2005, 117, 4330; Angew. Chem., Int. Ed. 2005, 44, 4258.
- 6. (a) Seebach, D.; Beck, A. K. Angew. Chem. 1974, 86, 859; Angew. Chem., Int. Ed. Engl. 1974, 13, 806; (b) Dumont, W.; Krief, A. Angew. Chem. 1976, 88, 184; Angew. Chem., Int. Ed. Engl. 1976, 15, 161; (c) Krief, A.; Dumont, W.; Denis, J. N. J. Chem. Soc., Chem. Commun. 1985, 571.
- 7. Typical procedure: A dry and argon-flushed 25 mL Schlenk flask, equipped with a magnetic stirring bar and a septum, was charged with a solution of phenylselenol (157 mg, 1.00 mmol) in dry THF (3 mL). After cooling to  $-78$  °C, i-PrMgCl (1.88 mL, 2.01 equiv, 1.07 M in THF) was added dropwise and stirred for 30 min. 2-Iodo-3-methoxyphenyl 4-chlorobenzenesulfonate (1b) (424 mg, 1.00 mmol) dissolved in dry THF (2 mL) was then added. The resulting mixture was stirred vigorously for 30 min at  $-78$  °C and immediately warmed to room temperature. After 1 h, the reaction mixture was cooled to  $-40$  °C and DMF (0.12 mL, 1.5 equiv) was added. The mixture was warmed to room temperature and stirred for 1 h. The reaction was quenched with satd aq NH<sub>4</sub>Cl solution, extracted with  $CH_2Cl_2$  $(3 \times 40 \text{ mL})$  and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After filtration, the solvent was evaporated in vacuo. Purification by flash chromatography (*n*-pentane/diethyl ether = 10:1) yielded 10b as a yellow solid (207 mg, 70%).
- 8. Selected data: Compound 10b: mp  $77.4-79.9$  °C. <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3, 25 \text{ }^{\circ}\text{C})$ :  $\delta = 10.65$  (s, 1H), 7.70–7.66 (m, 2H), 7.46–7.37 (m, 3H), 7.15 (t,  $\overline{3}J(\overline{H},H) = 8.2$  Hz, 1H), 6.71 (d, <sup>3</sup>J(H,H) = 8.2 Hz, 1H), 6.47 (d, <sup>3</sup>J(H,H) = 8.2 Hz,<br>1H), 3.91 (s, 3H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>, 25 °C):

 $\delta = 190.0, 164.0, 142.6, 137.4, 134.4, 129.6, 129.1, 128.6,$ 122.2, 120.9, 107.1, 55.8. MS (70 eV, EI): m/z (%): 294 (20), 293 (20), 292 (100) [M+], 291 (29), 290 (49), 289 (28), 288 (22), 215 (21), 214 (52), 212 (28), 135 (21), 134 (19), 77 (25), 76 (19). IR (KBr):  $v$  (cm<sup>-1</sup>) = 2888 (w), 1650 (s), 1579 (s), 1564 (vs), 1463 (s), 1437 (m), 1396 (m), 1295 (w), 1267 (vs), 1212 (w), 1187 (w), 1033 (m), 827 (w), 776 (w), 743 (w), 693 (w). HRMS for C<sub>14</sub>H<sub>12</sub>O<sub>2</sub>Se (292.0002): found: 292.0027. Compound 13a: mp  $64.5-65.4$  °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta = 8.23$  (d, <sup>4</sup>J(H,H) = 1.8 Hz, 1H), 7.82 (d,  $\frac{4}{4}$ J(H) = 1.8 Hz, 1H), 7.82 (d,  $J^4J(H,H) = 1.8$  Hz, 1H), 7.48–7.42 (m, 2H), 7.35–7.29 (m, 3H), 4.33 (q, <sup>3</sup>J(H,H) = 7.1 Hz, 2H), 2.85 (q, <sup>3</sup>J(H,H) = 7.1 Hz, 2H), 1.36 (t, <sup>3</sup>J(H,H) = 7.1 Hz, 3H), 1.26 (t, <sup>3</sup>J(H,H) = 7.1 Hz, 3H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta = 206.0, 164.0, 146.1, 145.9, 138.0, 133.6, 130.3,$ 129.7, 129.6, 129.3, 128.4, 94.3, 62.1, 37.0, 14.0, 7.6. MS  $(70 \text{ eV}, \text{EI})$ :  $m/z$  (%): 488 (60) [M<sup>+</sup>], 486 (29), 459 (59), 457 (26), 431 (30), 413 (26), 355 (23), 353 (100), 351 (51), 349<br>(20), 57 (20). IR (KBr):  $v \text{ (cm}^{-1)} = 2983 \text{ (w)}$ , 2938 (w), 1723 (vs), 1700 (s), 1558 (m), 1543 (w), 1438 (w), 1366 (w), 1279 (s), 1253 (s), 1196 (m), 1113 (m), 1020 (w), 948 (w), 868 (w), 785 (w), 765 (w), 742 (m), 722 (w), 692 (m). HRMS for  $C_{18}H_{17}IO_3$ Se (487.9388): found: 487.9415. Compound 16b: mp  $105.3-105.9$  °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta = 8.20{\text{-}}8.10$  (m, 2H), 7.57–7.45 (m, 4H), 7.38–7.17<br>(m, 7H), 4.46 (q, <sup>3</sup>J(H,H) = 7.1 Hz, 2H), 1.46 (t, 3<sub>J(H</sub>) = 7.1 H<sub>z</sub>, 3H), <sup>13</sup>C NMP (75 MHz, CDCL)  $J(H,H) = 7.1$  Hz, 3H).  $13^2C$  NMR (75 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta = 166.2, 146.0, 142.4, 134.5, 132.0, 131.9, 130.0,$ 129.9, 129.4, 129.3, 129.2, 129.0, 128.5, 127.8, 126.7, 60.8, 14.3. MS (70 eV, EI):  $m/z$  (%): 382 (100) [M<sup>+</sup>], 337 (15), 308 (12), 274 (7), 229 (63), 207 (7), 168 (4), 152 (31), 113 (3), 77 (3), 51 (1). IR (KBr):  $\nu$  (cm<sup>-1</sup>) = 2987 (m), 2906 (m), 1961 (w), 1709 (vs), 1609 (s), 1459 (m), 1440 (m), 1401 (m), 1275 (vs), 1178 (s), 1110 (s), 1027 (s), 858 (s), 746 (vs), 446 (m). HRMS for  $C_{21}H_{18}O_2$ Se (382.0472): found: 382.0449.